Hydrogen-Bonding as a Stereocontrolling Element in Free-Radical C-Allylation Reactions: Vicinal, Proximal, and Remote Asymmetric Induction in the Amino Acid Series

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Stereocontrolled bond-forming reactions by free-radical processes in acyclic molecules are currently among the more challenging areas in asymmetric synthesis.¹ Recent advances in this area have generally focused on the utilization of derivatives of carboxylic acids with appropriate chiral auxiliaries² and on exploiting the presence of $\hat{\beta}$ -alkoxy substituents and other resident groups.³ A model for the transition state in such reactions involving 1,2-asymmetric induction has been proposed by Hart and Krishnamurthy⁴ and further elaborated upon by Curran and co-workers.⁵ The general consensus is that, with few exceptions, the high diastereoselectivity is the result of a combination of minimized allylic 1,3-strain and torsional strain⁶ and of stereoelectronic effects.^{7,8} The notion that intramolecular H-bonding might play a role in the C-deuteration⁹ and C-allylation^{2d} of β -hydroxycarbalkoxy and amide α -radicals has been alluded to and experimentally tested with unrewarding⁴ or inconclusive results.5

Continuing our studies¹⁰ on the reactivity and stereochemical issues related to α -carbalkoxy radicals,¹¹ we report herein

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Scheme 1^a



^{*a*} Reaction conditions: (a) LDA, PhSeBr, THF, -78 °C; (b) allyltributylstannane, AIBN, $h\nu$, toluene or dichloromethane, -40 °C or -20 °C.

examples of remarkably high stereoselectivity in 1,2-induction^{2,3} and unprecedented 1,3-, 1,4-,¹² and 1,5-asymmetric inductions in the free-radical *C*-allylation of α -acyl radicals derived from a series of *N*-substituted acyclic amino acid derivatives (Scheme 1),¹³ by exploiting intramolecular *H*-bonding as a stereocontrolling element.

Treatment of the readily available α -selenophenyl or α -iodo esters or amides of β -N-substituted amino acid derivatives with allytributylstannane¹⁴ led to a quasi exclusive formation of the *anti-C*-allyl derivatives, regardless of the nature of the β -alkyl (aryl) substituent tested so far (Scheme 1, series A; Table 1, entries 1–3).¹⁵ Changing the β -substituent to a carbomethoxy group resulted in a complete reversal of selectivity, giving the *syn-C*-allylated product, while maintaining high stereoselectivity (Scheme 1, series B; Table 1, entries 4 and 5).¹⁵

Using DMSO instead of toluene as solvent resulted in an erosion or reversal of selectivity,¹⁵ thus demonstrating the role of the polarity of the solvent⁴ on the prevailing conformers in the transition state of these reactions and the dominant role that intramolecular H-bonding may play in controlling conformational mobility.

In an effort to study the influence of intramolecular H-bonding on the stereoselectivity of *C*-allylation, we extended the reaction to a series of γ -, δ -, and ω -amino acid derivatives (Scheme 1, series C, D). Unprecedented levels of proximal 1,3-induction were obtained with the *N*-trifluoroacetyl and *N*,*N*-dimethyl or *N*-methoxy-*N*-methyl amide derivatives (Table 1, entries 6–10). The selectivity was somewhat lower in the case where R₁ = Me and Ph (Table 1, entries 9 and 10). However, extending the distance between the electronically most favored hydrogenbonding partners (*N*-TFA and CONMe₂) led to surprisingly good 1,4-*anti*- and 1,5-*syn* selectivity (Table 1, entries 11 and 12; Scheme 1, series D, n = 2, 3, respectively).

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Table 1



^{*a*} Ratio determined by ¹H NMR of the crude sample. Entries 3, 9, 11, and 12 correspond to racemic products. ^{*b*} Isolated yield. ^{*c*} Stereochemistry determined by analogy to entry 3, and by NOE studies of the corresponding *N*-Boc-pyrrolidine derivative after cyclization with Hg(OAc)₂; see ref 15. ^{*d*} Stereochemistry determined by converting into *N*-Piv product, which was proved by X-ray single crystal analysis.^{*e*} Ratio determined by HPLC analysis. ^{*f*} Ratio for NMe₂ amide remained unchanged at 30 °C. ^{*s*} Stereochemistry correlated by chemical means; see ref 15. ^{*h*} The nitrogen atom was protected with Cbz instead of TFA. ^{*i*} Stereochemistry by analogy to the product in entry 8. ^{*j*} Anti/syn 86: 14, 64% for *N*,*N*-dimethyl derivative. ^{*k*} X-ray single crystal analysis. ^{*i*} Reaction done at 0 °C.



Figure 1. Proposed H-bonded radical intermediates favoring *anti* attack in 1,2-asymmetric induction (series A). Shaded lobe indicates preferred side of attack by allylstannane.

The quasi exclusive formation of *anti*-products in series A (1,2-induction) can be interpreted on the basis of the prevalence of ground state and transition state conformations, in which H-bonding plays a dominant role in favoring a pseudo sixmembered ring. Interestingly, the (2S)-anti isomer **A** (Figure 1), with a distinctive FT-IR H-bonding band at 3293 cm^{-1,16} reacted faster than the (2R)-syn isomer **B** (FT-IR 3410 cm⁻¹).

Finally, we demonstrate the versatility of the methodology

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^{*a*} Reaction conditions: (a) allyltributylstannane, AIBN, $h\nu$, toluene, -40 °C; (b) TBAF, THF; (c) peptide formation, PyBOP, *i*-Pr₂NEt, CH₃CN, with the corresponding amine salt (ref 15).

in iterative, and one-step, multiple *C*-allylation protocols leading to the tripeptide congener **2** (Scheme 2). Thus, *C*-allylation of the (2*S*)-tri- α -phenylseleno peptide derivative **1** gave a major product **2**, whose structure and stereochemistry was confirmed by an independent stepwise synthesis via amide coupling of enantiopure *C*-allyl precursors. Alternatively, *C*-allylation of **3**, amide formation, and iteration of the process on the α -phenylseleno dipeptide ester **4** also gave **2** (Scheme 2).^{15,17,18}

We have shown that H-bonding can be a strong stereocontrolling element in the free-radical *C*-allylation of a variety of acyclic amino acid derivatives. The diastereomerically pure or enriched products are versatile chirons for further manipulations¹⁹ and for the design of unnatural oligomeric amino acid motifs with potentially interesting three-dimensional arrays.²⁰ Results pertaining to these and related studies will be communicated in due course.

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Supporting Information Available: Physical constants, chemical correlations, experimental procedures, NMR and FT-IR spectra, and HPLC and X-ray analyses for the compounds described herein (36 pages). This material is contained in many libraries on microfiche, immediately follows this article in the microfilm version of the journal, can be ordered from the ACS, and can be downloaded from the Internet; see any current masthead page for ordering information and internet access instructions.

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